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EDITIONS: This book is available in paperback in 5.5" x 8.5" (portable size), 8.5" x 11" (large size), and as an eBook. This 5.5" x 8.5" edition is the most portable, while the details of the figures - including the periodic tables - are most clear in the large size and large print edition. However, the paperback editions are in black-and-white, whereas the eBooks are in color. OVERVIEW: This book focuses on fundamental chemistry concepts, such as understanding the periodic table of the elements and how chemical bonds are formed. No prior

knowledge of chemistry is assumed. The mathematical component involves only basic arithmetic. The content is much more conceptual than mathematical.

AUDIENCE: It is geared toward helping anyone - student or not - to understand the main ideas of chemistry. Both students and non-students may find it helpful to be able to focus on understanding the main concepts without the constant emphasis on computations that is generally found in chemistry lectures and textbooks. CONTENTS: (1) Understanding the organization of the periodic table, including trends and patterns. (2) Understanding ionic and covalent bonds and how they are formed, including the structure of valence electrons. (3) A set of rules to follow to speak

the language of chemistry fluently: How to name compounds when different types of compounds follow different naming schemes. (4)

Understanding chemical reactions, including how to balance them and a survey of important reactions. (5)

Understanding the three phases of matter: properties of matter, amorphous and crystalline solids, ideal gases, liquids, solutions, and acids/bases. (6)

Understanding atomic and nuclear structure and how it relates to chemistry. (7)

VERBAL ReACTiONS: A brief fun diversion from science for the verbal side of the brain, using symbols from chemistry's periodic table to make word puzzles.

ANSWERS: Every chapter includes self-check exercises to offer practice and help the reader check

his or her understanding. 100% of the exercises have answers at the back of the book. COPYRIGHT: Teachers who purchase one copy of this book or borrow one copy of this book from a library may reproduce selected pages for the purpose of teaching chemistry concepts to their own students.

Oswaal ICSE Question Bank Class 9 Physics, Chemistry, Math & Biology (Set of 4 Books) (For 2022-23 Exam) World Scientific

Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals. For over 35 years, this book has helped them master the chemistry skills they need to succeed. It provides them with clear and logical explanations of chemical concepts and problem solving. They'll learn how to apply concepts with the help of worked out examples. In addition, *Chemistry in Action* features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis.

CliffsStudySolver:

Chemistry Chemical formulas and names
Chemistry 2e Molecules with Silly Or Unusual Names

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dealing with an overview of chemistry, this book goes on exploring the concept of matter, energy, measurement, problem solving, atom, periodic table, and chemical bonding. These topics are followed by discussions on writing names and formulas of compounds; chemical formulas and the mole; chemical reactions; calculations based on equations; gases; and the properties of a liquid. The remaining chapters examine the solutions; acids; bases; salts; oxidation-reduction reactions; electrochemistry; chemical kinetics and equilibrium; and nuclear, organic, and biological chemistry. This study guide will be of great value to chemistry teachers and students.

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