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different “accounts” and can be Chemistry—Unit 3 Energy and Heating/Cooling Chemistry I Unit 3 Review Guide: “Energy and Electrons” Practice Questions and Problems 1. Energy is the capacity to do work. With reference to this definition, describe how you would demonstrate that each of the following has potential energy. (There is no one correct answer in these cases.)

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Quantitative Energy Problems Energy constants (H₂O) 334 J/g 2260 J/g 2.1 J/gC 4.18

Unit 3 Lab: Icy Hot - University of Kentucky

The total thermal energy in an object or substance The sun, fire pit, an oven, a Bunsen burner, etc. Joules (J)... Kilojoules (KJ)...

Calories (C)... Cannot be measured... Calculated using $Q=M$ (heat) C (specific heat)...

Heat The total thermal energy in an object or substance Examples of heat The sun, fire pit, an oven, a Bunsen burner, etc.

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Chemistry I Unit 3 Review Guide: “Energy and Electrons” Practice Questions and Problems 1. Energy is the capacity to do work.

With reference to this definition, describe how you would demonstrate that each of the following has potential energy. (There is no one correct answer in these cases.)

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