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Fuels - Higher and Lower Calorific Values

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O.P. Gupta is the author of Elements of Fuels, Furnaces & Refractories

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Higher and lower calorific values (=heating values) for some common fuels - coke, oil, wood, hydrogen and others. The resulting energy value divided by grams of fuel burned gives the energy content (in J/g). The combustion process generates water vapor and certain techniques may be used to recover the quantity of heat contained in this water vapor by condensing it.

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Classification of fuels,

properties and tests, coal origins, carbonization and gasification. Other solid fuels. Liquid fuels-Types, testing, properties, Gaseous fuels, Introduction to nuclear fuels, Indian fuel deposits. Principles of theory of combustion, Combustion calculations, Waste heat utilization. Classification of furnaces operation.

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Section - D :Gaseous Fuels *Methane, wood gas, gobar gas, sewage gas and gas from underground gasification of coal *Natural gas, liquefied petroleum gas (LPG) and refinery gases *Producer gas and water gas *Blast furnace gas, coke oven gas and L.D. converter gas/steel plant fuels *Coal gas from coal gasification processes *Oil gas from oil gasification processes.

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Classification and Description

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Subsequent chapters focus on the thermodynamics, physics, chemistry, and kinetics of combustion of fuels; the burner design; the heat transfer and flow of gases through furnaces and flues; and ways of controlling energy supply rates and temperatures. Most furnaces use liquid fuel, gaseous fuel or electricity as energy input. Induction and arc furnaces use electricity to melt steel and cast iron. Melting furnaces for nonferrous materials use fuel oil. Oil-fired furnaces mostly use furnace oil, especially for reheating and heat treatment of materials.

FURNACES AND REFRATORIES

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Furnaces ...

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